



Saikachinoside A, a novel 3-prenylated isoguanine glucoside from seeds of *Gleditsia japonica*

Tadashi Kajimoto, Nobuwa Aoki, Emi Ohta, Yasushi Kawai, Shinji Ohta *

Nagahama Institute of Bio-Science and Technology, 1266 Tamura-cho, Nagahama, Shiga 526-0829, Japan

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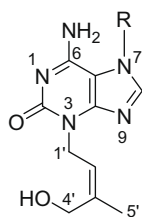
Leguminosae

ABSTRACT

A new isoguanine glucoside designated saikachinoside A has been isolated from the seeds of *Gleditsia japonica*. The structure was determined as 7-β-D-glucopyranosyl-3-[(Z)-4-hydroxy-3-methyl-2-butenyl]isoguanine by interpretation of the spectroscopic data and was confirmed by X-ray crystallographic analysis. Saikachinoside A accelerated acetylcholinesterase activity.

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The naturally occurring isopentenyl derivatives of purines presently known consist of triacanthine¹ and isoprenoid cytokinins^{2,3} such as N⁶-isopentenyladenine,⁴ *trans*-zeatin,⁵ *cis*-zeatin,^{6,7} dihydrozeatin,⁸ and their sugar conjugates.^{9–11} In the course of our continuing search for biologically active compounds from plants,^{12,13} we isolated a new 3-prenylated isoguanine¹⁴ glucoside designated saikachinoside A (**1**) from the MeOH extract of seeds of Japanese honey locust, *Gleditsia japonica* Miquel (Leguminosae) which have been used as diuretics and expectorants in oriental traditional medicine.¹⁵ This report describes the purification and the structure elucidation of **1**.



- 1 R = β-D-Glucopyranosyl
2 R = H

The pods of *G. japonica* were collected in Shiga Prefecture, Japan in March 2009. Voucher specimens are kept at Nagahama Institute of Bio-Science and Technology. The seeds (80 g, wet weight) were cut into small pieces and extracted with MeOH. The concentrated

MeOH extract was suspended in water and partitioned successively with hexane and EtOAc. The aqueous layer was lyophilized to yield syrup. The water-soluble fraction (2.4 g) was separated repeatedly on an ODS column employing MeOH in H₂O gradient mixtures to afford **1** (27 mg, 0.03% wet weight) as colorless plates, mp 205–208 °C, [α]_D²⁵ +9.0 (c 1.1, H₂O).

Saikachinoside A (**1**) exhibited pseudomolecular ion peaks at *m/z* 398 and 396 corresponding to [M+H]⁺ and [M–H][–] in the (+)- and (–)-ESIMS, respectively. The molecular formula of **1** was established to be C₁₆H₂₃N₅O₇ on the basis of high-resolution ESITOFMS data (*m/z* 398.1677 [M+H]⁺, Δ +0.1 mmu). The IR spectrum displayed absorption bands at 3600–2600 (OH, NH, and CH), 1635, 1626, and 1577 cm^{–1} (C=O, C=N, and C=C). The UV (H₂O) absorption maxima at 246 (log ε 4.03) and 289 nm (log ε 4.00) suggested the presence of a 3,7-disubstituted isoguanine skeleton.^{16,17} The presence of the isoguanine unit was supported by the ¹H and ¹³C NMR (DMSO-*d*₆) chemical shifts [δ_H 8.06 (1H, s, H-8), δ_C 154.5 (C-2), 152.6 (C-4), 102.4 (C-5), 153.1 (C-6), 142.5 (C-8)]. Inspection of the ¹H and ¹³C NMR spectra (Table 1) together with DEPT and HMQC spectral data revealed the presence of an olefinic methyl, three hetero atom bearing methylenes, five oxygenated methines, an olefinic methine, and a quaternary olefinic carbon in addition to the isoguanine unit. The interpretation of ¹H vicinal *J*-coupling constants and ¹H–¹H COSY spectrum revealed the presence of a monosaccharide unit which was supported by the observation of the fragment ion peak at *m/z* 234 [M–163][–] in the negative ion MS/MS spectrum. The large vicinal coupling constants of H-1'' to H-5'' implied that these protons are all axial, indicating the monosaccharide unit to be β-glucopyranosyl. The HMBC (Fig. 1) from H-1'' to C-5 and C-8 and from H-8 to C-1'' indicated the attachment of the β-glucopyranosyl unit at N-7 of the isoguanine unit. The find-

* Corresponding author. Tel.: +81 749 64 8101; fax: +81 749 64 8140.
E-mail address: s_ohta@nagahama-i-bio.ac.jp (S. Ohta).

Table 1
NMR spectral data of **1** in DMSO-*d*₆ at 40 °C^a

No	δ_C	δ_H multiplicity (<i>J</i> in Hz)	HMBC
2	154.5 s		
4	152.6 s		
5	102.4 s		
6	153.1 s		
8	142.5 d	8.06 s	C-4, C-5, C-1''
1'	39.3 t	4.59 d (2H, 7.0)	C-2, C-4, C-2', C-3'
2'	120.7 d	5.26 br t (7.0)	C-4', C-5'
3'	139.0 s		
4'	59.9 t	4.11 d (2H, 5.5)	C-2', C-3', C-5'
5'	21.2 q	1.68 br s (3H)	C-2', C-3', C-4'
1''	86.2 d	5.40 d (8.4)	C-5, C-8, C-2'', C-3''
2''	72.3 d	3.33 br dd (9.5, 8.4)	C-3''
3''	76.2 d	3.35 br t (9.5)	C-2''
4''	68.0 d	3.46 br t (9.5)	C-3'', C-6''
5''	79.2 d	3.51 dt (9.5, 2.2t)	C-1'', C-3'', C-4''
6''	58.9 d	3.67 br s (2H)	C-5''
6-NH ₂		7.08 br s (2H)	
4'-OH		4.72 t (5.5)	C-3', C-4'
2''-OH		5.44 br s	
3''-OH		5.21 br s	
4''-OH		5.15 br s	
6''-OH		4.90 br s	C-5''

^a The ¹H and ¹³C NMR were measured at 400 and 100 MHz, respectively. Chemical shifts were referenced to solvent (δ_H 2.49 and δ_C 39.5).

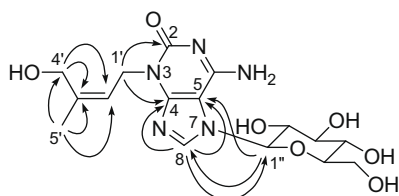


Figure 1. Key HMBC correlations observed for **1**.

ings were supported by an NOE experiment in which the enhancement of 5% on the anomeric proton (H-1'') was observed upon irradiation of H-8 (Fig. 2).

The presence of a 4-hydroxyisopentenyl unit was established on the basis of the HMBC (H₂-1'/C-2' and C-3'; H-2'/C-4' and C-5'; H₂-4'/C-2', C-3', and C-5'; H₃-5'/C-2', C-3', and C-4'). The HMBC from H-1' to C-2 and C-4 indicated the attachment of the 4-hydroxyisopentenyl unit at N-3 of the isoguanine unit. The NMR chemical shifts of H₃-5' (δ_H 1.68) and C-5' (δ_C 21.2) suggested the geometry of Δ^2 to be *Z* [2-Me of (2*Z*,6*E*)-2,6-dimethyl-2,6-octadiene-1,8-diol: δ_H 1.79 and δ_C 21.1¹⁸] and not to be *E* [2-Me of (*E*)-4-phenyl-2-methyl-2-buten-1-ol: δ_H 1.80 and δ_C 13.8¹⁹]. The geometry was supported by difference NOE experiments. Enhancements of the oxygenated methylene protons (H₂-4') and the olefinic proton (H-2') upon irradiations of the allyl protons (H₂-1') and the methyl protons (H₃-5'), respectively, were observed as shown in Figure 2. The structure of **1** was confirmed by X-ray crystallographic analysis (Fig. 3).²⁰

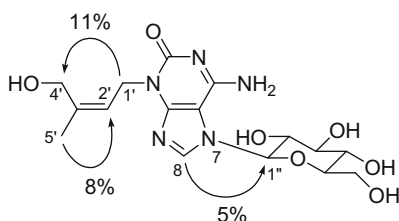


Figure 2. Key NOEs observed for **1**.

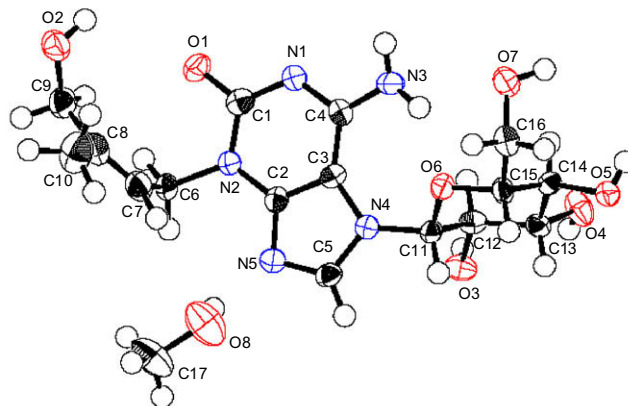


Figure 3. Perspective view of the crystal structure of **1**.

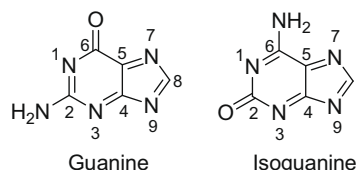
Hydrolysis of **1** with 1 M HCl afforded aglycone **2** and D-glucose.²¹ Consequently, the structure of saikachinoside A was elucidated as 7- β -D-glucopyranosyl-3-[(*Z*)-4-hydroxy-3-methyl-2-butenyl]isoguanine (**1**).

Although several naturally occurring prenylated purines such as isoprenoid cytokinins have been reported, most of them are *N*⁶-prenylated adenine derivatives.^{2–8} There are very few, if any, compounds having a 3-prenylated purine skeleton.^{1,22}

Saikachinoside A (**1**) was evaluated using electric eel acetylcholinesterase (AChE) following the colorimetric method of Ellman et al.²³ with some modifications.^{24–26} The addition of 2.5×10^{-4} M of **1** led to 1.3-fold increase in the AChE activity. It has been reported that several compounds, such as polyamines, activate AChE by allosteric mechanism.^{27,28}

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20. Crystal data for **1**: C₁₆H₂₃N₅O₇·CH₃OH, *M* = 429.43, orthorhombic, space group *P*2₁2₁2₁ (#19), *Z* = 4, *a* = 8.358 (3), *b* = 13.532 (4), *c* = 17.820 (7) Å, *V* = 2015.6 (12) Å³, *F*(0 0 0) = 912, μ(Mo-Kα) = 1.130 cm⁻¹, *D*_c = 1.415 g cm⁻³, *T* = 93 K. The reflection data were collected on a Rigaku RAXIS RAPID imaging plate area detector with graphite-monochromated Mo-Kα radiation (λ = 0.71075 Å). A total of 4609 independent reflections were collected of which 3192 were considered to be observed [*I* > 2.0σ(*I*)]. The structure was solved by direct methods and expanded using Fourier techniques.²⁹ The non-hydrogen atoms were refined anisotropically by full-matrix least-squares refinement.³⁰ Hydrogen atoms were refined isotropically. The structure was finally refined to *R* = 0.0330 (*wR*₂ = 0.0699). Crystallographic data for the structure in this Letter has been deposited with the Cambridge Crystallographic Data Centre as supplementary publication number CCDC 756695. Copies of the data can be obtained, free of charge, via www.ccdc.cam.ac.uk (or from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge, CB2 1EZ, UK; Fax: +44 1223 336 033; or e-mail: deposit@ccdc.cam.ac.uk).
21. Treatment of **1** (6 mg) with 1 M HCl at 90 °C overnight gave a reaction mixture. The mixture, after neutralization, was subjected to a short ODS column chromatography to afford aglycone **2** (2 mg) {ESIMS *m/z* 236 [M+H]⁺} and a monosaccharide (2 mg). The monosaccharide was identified to be D-glucose on the basis of co-HPLC (Wakosil 5NH₂ column, 75%CH₃CN-H₂O, RI detector) with an authentic sample and the specific optical rotation {[α]_D +45 (c 0.1, H₂O)}.
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26. The enzymatic activity was measured in 100 mM phosphate buffer, pH 8.0, at 37 °C, using acetylthiocholine (ATC, 0.5 mM) and 5,5'-dithio-bis(2-nitrobenzoic) acid (DTNB, 0.25 mM). The initial rate of substrate hydrolysis was determined using a microplate reader at 415 nm. Physostigmine was used as the reference compound.
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